



# AMINOSALICYLATE SODIUM DIHYDRATE

(AMINOSALICYLATUM NATRICUM DIHYDRICUS)

## Draft proposal for inclusion for *The International Pharmacopoeia*

(15 November 2024)

*DRAFT FOR COMMENTS*

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For any technical queries, please contact **Dr Herbert Schmidt**, Technical Officer, Norms and Standards for Pharmaceuticals, Technical Standards and Specifications ([schmidt@who.int](mailto:schmidt@who.int)), with a copy to Ms Sinéad Jones ([jonesi@who.int](mailto:jonesi@who.int), [nsp@who.int](mailto:nsp@who.int)).

Comments should be submitted through the online platform on or by **15 January 2025**. Please note that only comments received by this deadline will be considered for the preparation of this document.

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SCHEDULE FOR THE ADOPTION PROCESS OF DOCUMENT QAS/20.844:

**AMINOSALICYLATE SODIUM DIHYDRATE**  
**(AMINOSALICYLATUM NATRICUM DIHYDRICUS)**

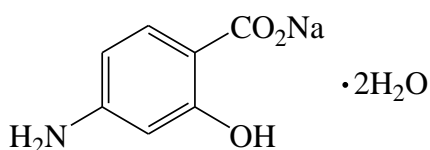
Description	Date
Draft monograph received from Collaborating Centre.	March 2020
Discussion at the consultation on Screening Technologies, Laboratory Tools and Pharmacopoeial Specifications for Medicines.	May 2020
Discussion at the consultation on Quality Control and Pharmacopoeial Specifications of Medicines	April 2023
Discussion at the consultation on Quality Control and Pharmacopoeial Specifications of Medicines	May 2024
Draft monograph sent out for public consultation.	November 2024 – January 2025
Presentation to the 59 <sup>th</sup> WHO Expert Committee on Specifications for Pharmaceutical Preparations.	October 2025
Further follow-up action as required.	

**AMINOSALICYLATE SODIUM DIHYDRATE**  
**(AMINOSALICYLATUM NATRICUM DIHYDRICUS)**

**Molecular formula.**  $C_7H_6NNaO_3 \cdot 2H_2O$

**Relative molecular mass.** 211.2

**Graphic formula.**



**Chemical name.** Sodium 4-amino-2-hydroxybenzoate dihydrate;

**CAS Registry Number.** 6018-19-5.

**Description.** White or almost white, crystalline powder or crystals.

**Solubility.** Freely soluble in water R; sparingly soluble in ethanol (~750 g/L) TS,  
practically insoluble in dichloromethane R.

**Category.** Antituberculosis.

**Storage.** Aminosalicylate sodium should be kept in a tightly closed container, protected  
from light.

**Additional information.** Aminosalicylate sodium is slightly hygroscopic.

**Requirements**

**Definition.** Aminosalicylate sodium contains not less than 99.0% and not more than  
101.0% of  $C_7H_6NNaO_3$ , calculated with reference to the dried substance.

### 63 Identity tests

- 64 • Either tests A and G, tests B and G, or any two of tests C, D, E or E, together with  
65 test G, may be applied.

66 A. Carry out the examination as described under *1.7 Spectrophotometry in the*  
67 *infrared region*. The infrared absorption spectrum is concordant with the spectrum  
68 obtained from aminosalicylate sodium RS or with the reference spectrum of  
69 aminosalicylate sodium.

70 B. Carry out the test as described under *1.14.1 Chromatography*, High-performance  
71 liquid chromatography, using the conditions and solutions given under “Identity  
72 test C” with the following modification: as the detector, use a diode array detector  
73 to record the UV spectrum of the principal peak in each chromatogram in the  
74 range of 220 nm to 400 nm.

75 The retention time and the UV spectrum of the principal peak in the  
76 chromatogram obtained with solution (A) corresponds to the retention time and  
77 the UV spectrum of the peak due to aminosalicylate in the chromatogram obtained  
78 with solution (B).

79 C. Carry out the test as described under *1.14.1 Chromatography*, High-performance  
80 liquid chromatography, using the conditions given under “Related substances”.  
81 Prepare the following solutions. For solution (A), dilute 1.0 mL of solution (1) as  
82 described under “Related substances” to 10.0 mL with water R. For solution (B),  
83 dissolve 10 mg of the aminosalicylate sodium RS in water R and dilute to 100.0  
84 mL with the same solvent.

85 The retention time of the principal peak in the chromatogram obtained with  
86 solution (A) corresponds to the retention time of the peak due to aminosalicylate  
87 in the chromatogram obtained with solution (B).

D. Carry out as described under *1.6 Spectrophotometry in the visible and ultraviolet regions*. Transfer 25 mg of the test substance to a 50 mL volumetric flask, add 3 mL of sodium hydroxide (~40 g/L) TS to dissolve the substance and dilute with water R to volume. Transfer 5 mL of this solution to a 250 mL volumetric flask, add 12.5 mL of phosphate buffer, pH 7.0 (0.05 mol/L) TS and dilute with water R to volume. The absorption spectrum of the solution, when observed between 220 nm and 400 nm, exhibits maxima at 265 and 299 nm.

E. Carry out the test as described under *1.14.1 Chromatography, Thin-layer chromatography*, using silica gel R6 as the coating substance and a mixture of 50 volumes of ethyl acetate R, 1 volume of ethanol (~750 g/L) TS and 1 volume of anhydrous formic acid R as the mobile phase. Apply separately to the plate 5 µL of each of the following two solutions. For solution (A), add 5 mL of ethanol (~750 g/L) TS to 10 mg of the test substance, shake and filter. For solution (B), use a solution containing 2 mg of aminosalicylate sodium RS per mL of ethanol (~750 g/L) TS. After removing the plate from the chromatographic chamber, allow it to dry in air. Examine the chromatogram in ultraviolet light (254 nm).

The principal spot obtained with solution (A) corresponds in position, appearance and intensity to that obtained with solution (B).

F. 0.05 g of the test substance yields the reaction described for the identification of primary aromatic amines under *2.1 General identification tests*, producing an intense orange or red precipitate.

G. 0.01 g of the test substance yields reaction A described for the identification of sodium under *2.1 General identification tests*.

**pH value (1.13).** pH of a freshly prepared 20 mg per mL solution of the test substance in carbon-dioxide-free water R, 6.5-8.5.

**Clarity and colour of solution.** A freshly prepared solution of 2.5 g of the test substance in 25 mL of water R is clear and not more intensely colored than reference solution B<sub>5</sub>, when compared as described under *1.11.2 Degree of coloration of liquids*, Method II.

**Loss on drying.** Dry to constant weight at 105 °C; it loses not less than 160 mg per g and not more than 180 mg per g.

**Heavy metals.** Use 1.0 g for the preparation of the test solution as described under *2.2.3. Limit test for heavy metals*, Procedure 3; determine the heavy metals content according to method A; not more than 10 µg/g.

**Related substances.** Prepare the solutions and the mobile phase freshly using low-actinic glassware. Carry out the test as described under *1.14.1 Chromatography*, High-performance liquid chromatography, using a stainless steel column (25.0 cm × 4.6 mm) packed with base-deactivated and end-capped particles of silica gel, the surface of which has been modified with chemically-bonded octylsilyl groups (5 µm)<sup>1</sup>.

Use the following conditions for gradient elution:

**Mobile phase A:** dissolve 2.2 g of perchloric acid (~1170 g/L) TS and 1.0 g of phosphoric acid (~1440 g/L) TS in water R and dilute to 1000 mL with the same solvent;

**Mobile phase B:** dissolve 1.7 g of perchloric acid (~1170 g/L) TS and 1.0 g of phosphoric acid (~1440 g/L) TS in acetonitrile R and dilute to 1000 mL with the same solvent.

Time (minutes)	Mobile phase A (% v/v)	Mobile phase B (% v/v)	Comments
0–15	100	0	Isocratic
15–30	100 to 40	0 to 60	Linear gradient

<sup>1</sup> A Hypersil BDS C8column has been found suitable.

30–31	40 to 100	60 to 0	Return to initial composition
31–45	100	0	Re-equilibration

133 Operate with a flow rate of 1.25 mL per minute. As a detector, use an ultraviolet  
134 spectrophotometer at a wavelength of about 220 nm.

135 Prepare the following solutions using water R as the diluent.

136 For solution (1), dissolve 50 mg of the test substance and dilute to 50.0 mL.

137 For solution (2), dilute 10.0 mL of solution (1) to 100.0 mL. Dilute 2.0 mL of this  
138 solution to 100.0 mL.

139 For solution (3), dilute 3.0 mL of solution (2) to 20.0 mL.

140 For solution (4), dissolve 5.0 mg of 3-aminophenol R (impurity A) and 5 mg of  
141 mesalazine R (impurity B) and dilute to 100.0 mL.

142 For solution (5), dilute 5.0 mL of solution (4) to 100.0 mL.

143 Inject 10 µL each of solutions (1), (2), (3) and (5).

144 Use the chromatogram obtained with solution (5) to identify the peak due to impurity A  
145 and B. The impurities are eluted, if present, at the following relative retention with  
146 reference to aminosaliclate (retention time about 17 minutes): impurity A about 0.31  
147 and impurity B about 0.39.

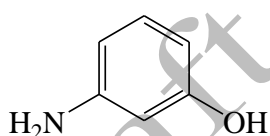
148 The test is not valid unless, in the chromatogram obtained with solution (5), the  
149 resolution between the peaks due to impurity A and impurity B is at least 4.0. Also, the  
150 test is not valid unless, in the chromatogram obtained with solution (3), the signal-to-  
151 noise ratio of the peak due to aminosaliclate is at least 10.

152 In the chromatogram obtained with solution (1):

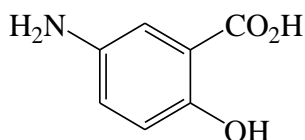
- the area of any peak corresponding to impurity A, when multiplied by a correction factor of 0.62, is not greater than 0.75 times the area of the peak due to aminosaliclate in the chromatogram obtained with solution (2) (0.15%);
- the area of any other impurity peak is not greater than 0.25 times the area of the peak due to aminosaliclate in the chromatogram obtained with solution (2) (0.05%).
- The sum of the areas of all impurity peaks is not greater than the area of the peak due to aminosaliclate in the chromatogram obtained with solution (2) (0.2 %). Disregard any peak with an area of less than the area of the peak due to aminosaliclate in the chromatogram obtained with solution (3) (0.03%).

**Assay.** Carry out the assay as described under 2.7 *Nitrite titration*. Dissolve 0.150 g of the test substance in 20 mL of water R. Add 10 mL of a 500 g per L solution of sodium bromide R and 25 mL of glacial acetic acid R. Add 5.0 mL of sodium nitrite (0.1 mol/L) VS rapidly and continue the titration with the same titrant, determining the endpoint potentiometrically. Each mL of sodium nitrite (0.1 mol/L) VS is equivalent to 17.52 mg of  $C_7H_6NNaO_3$ .

### Impurities



- A. 3-Aminophenol (m-Aminophenol) (synthesis-related impurity, degradation product),



- B. 5-Amino-2-hydroxybenzoic acid (mesalazine)(synthesis-related impurity).



176 **Reagents to be established**

177 **Phosphate buffer, pH 7.0 (0.05 mol/L) TS**

178 *Procedure.* Dissolve 0.681 g of potassium dihydrogen phosphate R in 100.0 mL of  
179 water. Adjust the pH using a 7,10 g/L solution of anhydrous disodium hydrogen  
180 phosphate R.

181 **Reagents to be established**

182 **3-Aminophenol R**

183  $C_6H_7NO$

184 *Contains not less than 97.0% of  $C_6H_7NO$*

185 *Molecular weight.* 109.1

186 *Description.* Pale yellowish-brown crystals, sparingly soluble in water.

187 *Melting point.* About 122 °C.

188 **Mesalazine R**

189  $C_7H_7NO_3$

190 *Molecular weight.* 153.1

191 *Description.* Almost white or light grey or light pink powder or crystals.

192 *Melting point.* 275 °C to 280 °C, with decomposition.

193 **Reference substances to be established**

194 **Aminosalicylate sodium RS**

195 New ICRS to be established.

196 \*\*\*